19950524 006

REPORT DOCUMENTATION PAGE

Form Approved
OMB No. 0704-0188

Public reporting burden for this collection of information is estimated to average 1 hour per response, including the time for reviewing instructions, searching existing data sources, gathering and maintaining the data needed, and completing and reviewing the collection of information. Send comments regarding this burden estimate or any other aspect of this collection of information, including suggestions for reducing this burden, to Washington Headquarters Services, Directorate for Information Operations and Reports, 1215 Jefferson Davis Highway, Suite 1204, Arlington, VA 22202-4302, and to the office of Management and Budget, Paperwork Reduction Project (0704-0188), Washington, DC 20503.

1. AGENCY USE ONLY (Leave blank)	2. REPORT DATE 05/18/95	3. REPORT TYPE AND Final Report 7	7/7/955-6/36/994
4. TITLE AND SUBTITLE			5. FUNDING NUMBERS
Block Copolymer Ionome	rs		N00014-56-J-0011
6. AUTHOR(S)			
R. A. Weiss			
7. PERFORMING ORGANIZATION NAME Institute of Materials Scie University of Connecticut 97 N. Eagleville Road Storrs, CT 06269-3136			8. PERFORMING ORGANIZATION REPORT NUMBER
9. SPONSORING/MONITORING AGENCY Office of Naval Research 800 N. Quincy Street Arlington, VA 22217-5000			10. SPONSORING / MONITORING AGENCY REPORT NUMBER
11. SUPPLEMENTARY NOTES	SELEC MAY 2 6	1995	
12a. DISTRIBUTION / AVAILABILITY STAT Distribution Unlimited	EMENT G		12b. DISTRIBUTION CODE
			DTIC QUALITY INSPECTED
13. ABSTRACT (Maximum 200 words)			

Three types of block copolymers were used to prepare block copolymer ionomers: (1) a triblock copolymer of styrene and ethylene-co-butylene (SEBS) obtained commercially and diblock and triblock copolymers of styrene and ethylene-co-propylene (SEP and SEPS) that were synthesized by anionic polymerization of styrene and isoprene followed by hydrogenation of the polyisoprene block. The styrene content of the SEBS was 30% (wt); the styrene content of the SEP and SEPS was varied from 10-50% to obtain spherical, cylindrical or lamellar microstructures. The styrene blocks were sulfonated to 0-18 mol% with acetyl sulfate, and the free acid derivative and the Na⁺ and Zn²⁺ salts were prepared. The block copolymer ionomers exhibited two different levels of microphase separation: (1) a block microstructure with a characteristic dimension of the order of 10-20 nm and an ionic microstructure of the order of 2-4 nm that was located within the styrene block microdomains. The effects of sulfonation, nature of the cation, temperature and sample preparation on the different microstructures and the properties of the polymers were studied by dynamic mechanical analysis, transmission electron microscopy, and small angle x-ray scattering. The effect of adding homopolymers that are miscible with the ionic phase was also studied.

14. SUBJECT TERMS	15. NUMBER OF PAGES		
Block Copolymers, Io	16. PRICE CODE		
17. SECURITY CLASSIFICATION OF REPORT Unclassified	18. SECURITY CLASSIFICATION OF THIS PAGE Unclassified	19. SECURITY CLASSIFICATION OF ABSTRACT Unclassified	20. LIMITATION OF ABSTRACT UL

GENERAL INSTRUCTIONS FOR COMPLETING SF 298

The Report Documentation Page (RDP) is used in announcing and cataloging reports. It is important that this information be consistent with the rest of the report, particularly the cover and title page. Instructions for filling in each block of the form follow. It is important to stay within the lines to meet optical scanning requirements.

- Block 1. Agency Use Only (Leave blank).
- Block 2. Report Date. Full publication date including day, month, and year, if available (e.g. 1 Jan 88). Must cite at least the year.
- Block 3. Type of Report and Dates Covered. State whether report is interim, final, etc. If applicable, enter inclusive report dates (e.g. 10 Jun 87 30 Jun 88).
- Block 4. <u>Title and Subtitle</u>. A title is taken from the part of the report that provides the most meaningful and complete information. When a report is prepared in more than one volume, repeat the primary title, add volume number, and include subtitle for the specific volume. On classified documents enter the title classification in parentheses.
- Block 5. Funding Numbers. To include contract and grant numbers; may include program element number(s), project number(s), task number(s), and work unit number(s). Use the following labels:

C - Contract G - Grant PR - Project
TA - Task

PE - Program Element TA - Task

WU - Work Unit

Accession No.

- Block 6. <u>Author(s)</u>. Name(s) of person(s) responsible for writing the report, performing the research, or credited with the content of the report. If editor or compiler, this should follow the name(s).
- Block 7. <u>Performing Organization Name(s) and</u> Address(es). Self-explanatory.
- Block 8. <u>Performing Organization Report</u>
 <u>Number</u>. Enter the unique alphanumeric report
 number(s) assigned by the organization
 performing the report.
- **Block 9.** Sponsoring/Monitoring Agency Name(s) and Address(es). Self-explanatory.
- **Block 10.** <u>Sponsoring/Monitoring Agency</u> Report Number. (*If known*)
- Block 11. Supplementary Notes. Enter information not included elsewhere such as: Prepared in cooperation with...; Trans. of...; To be published in.... When a report is revised, include a statement whether the new report supersedes or supplements the older report.

Block 12a. <u>Distribution/Availability Statement</u>. Denotes public availability or limitations. Cite any availability to the public. Enter additional limitations or special markings in all capitals (e.g. NOFORN, REL, ITAR).

DOD - See DoDD 5230.24, "Distribution Statements on Technical Documents."

DOE - See authorities.

NASA - See Handbook NHB 2200.2.

NTIS - Leave blank.

Block 12b. Distribution Code.

DOD - Leave blank.

 DOE - Enter DOE distribution categories from the Standard Distribution for Unclassified Scientific and Technical Reports.

NASA - Leave blank. NTIS - Leave blank.

- **Block 13.** Abstract. Include a brief (Maximum 200 words) factual summary of the most significant information contained in the report.
- **Block 14.** Subject Terms. Keywords or phrases identifying major subjects in the report.
- **Block 15.** <u>Number of Pages</u>. Enter the total number of pages.
- **Block 16.** Price Code. Enter appropriate price code (NTIS only).
- Blocks 17. 19. Security Classifications. Self-explanatory. Enter U.S. Security Classification in accordance with U.S. Security Regulations (i.e., UNCLASSIFIED). If form contains classified information, stamp classification on the top and bottom of the page.
- Block 20. <u>Limitation of Abstract</u>. This block must be completed to assign a limitation to the abstract. Enter either UL (unlimited) or SAR (same as report). An entry in this block is necessary if the abstract is to be limited. If blank, the abstract is assumed to be unlimited.

FINAL REPORT

Grant # N00014-56-J-0011

R&T Code: uri5302

PRINCIPAL INVESTIGATOR: R. A. Weiss

INSTITUTION: The University of Connecticut

GRANT TITLE: Synthesis and Characterization of Block Copolymer Ionomers with

Coexisting Dual Microphase Separated Morphologies

AWARD PERIOD: 1 July 1991 -- 30 June 1994

OBJECTIVE: To synthesize and characterize the structure and properties of block

copolymer ionomers.

APPROACH: Hydrogenated di- and triblock copolymers of styrene and a diene were either obtained from a commercial source or synthesized by anionic polymerization followed by hydrogenation of the diene block. Three types of copolymers were obtained: (1) triblock copolymers of styrene/ethylene-co-butylene/styrene (SEBS); (2) diblock copolymers of styrene/ethylene-co-propylene (SEP); and (3) triblock copolymers of styrene/ethylene-co-propylene/styrene (SEPS). The styrene content of the SEBS was 30%(wt); the styrene content of the SEP and SEPS was varied from 10-50% to accomplish block microstructures ranging from spherical to cylindrical to lamellar domains. The styrene blocks were lightly sulfonated with acetyl sulfate following the procedure of Makowski, Lundberg and Singhal (U.S. Pat. 3,870,841).

X

Codes

Sulfonate concentration varied from 0-18 mol% of the styrene and the free acid derivative and the

Na⁺ and Zn²⁺ salts were studied. The block copolymer ionomers exhibited two different levels

of microphase separation: (1) a block microstructure with a characteristic dimension of the order

Avair and or

Dist Avair and o Special

A-1

styrene block microdomains. The effects of sulfonation, nature of the cation, temperature and sample preparation on the different microstructures and the properties of the polymers were studied by dynamic mechanical analysis, transmission electron microscopy, and small angle x-ray scattering. The effect of adding homopolymers that are miscible with the ionic phase was also studied.

A. WORK ACCOMPLISHED [The number in brackets refer to publications in section B] Block Copolymer Synthesis [2,9,11,12]: SEBS block copolymers were obtained from Shell Development Corp. SEP and SEPS block copolymers were synthesized by anionic polymerization of styrene and isoprene followed by hydrogenation of the isoprene block. A summary of the latter diblock and triblock copolymers is given in Table 1, which includes the composition, molecular weight (all had narrow molecular weight distributions), the microstructure determined by small angle x-ray scattering (SAXS) and transmission electron microscopy (TEM), the interdomain spacing and the order-disorder transition temperature determined by dynamic mechanical analysis (DMTA).

The block copolymers were sulfonated using the procedure of Makowski, Lundberg and Singhal (U.S. Pat. 3,870,841). The sulfonating reagent was acetyl sulfate, and sulfonations were performed in homogenous solutions using dichloroethane. Sulfonate concentration varied from 0-18 mol% of the styrene and the free acid derivative and the Na⁺ and Zn²⁺ salts were studied.

PS, the morphology changed from body-centered cubic packed spherical domains of PS to hexagonal packed PS cylinders to alternating PS and PEP lamella. The ODT transition was determined by DMTA, and for a fixed composition and total polymer molecular weight, the $T_{\rm ODT}$ for the diblocks were about twice that of the triblocks. Partial sulfonation of the PS blocks inhibited the formation of the block microstructure, but did not change the texture. The inhibition to the development of order increased as the strength of the dipolar interactions increased, i.e., in the order of cation H < Zn < Na. The compression molded SEP50 sample exhibited two anomalies: (1) a rheological transition between T_g and $T_{\rm ODT}$ that may represent an order-order transition and (2) a mixed morphology of lamella and perforated lamella microstructures as observed by TEM.

Thermally-Induced Transitions [10]: Thermally-induced morphology changes of sulfonated SEBS were investigated using time-resolved SAXS measurements. At elevated temperature, order-order and/or order-disorder transitions of the block microstructure were observed following dissociation of the ionic aggregates for the free acid and zinc salt derivatives. The critical temperature for dissociation of the ionic phase was relatively insensitive to sulfonation level, but strongly dependent on the nature of the cation. The ionic dissociation temperature was ca. 175°C for the free acids, 235-250°C for the Zn-salts and was not observed for the Na-salts. Thermally induced transitions of the block microstructure were sensitive to the sulfonation level. At low sulfonation levels (e.g., 3 mol%), the order-disorder transition of the block microstructure was lower than the ionic dissociation temperature. As a result, the order-disorder transition was

sulfonate concentrations (9 mol%), an order-order transition from a spherical to a lamellar block microstructure occurred following dissociation of the ionic aggregates. This was also accompanied by phase mixing of the blocks, which indicated that the order-disorder transition was close to, but above, the ionic dissociation temperature. At the highest sulfonation studied (18 mol%), the order-disorder transition of the block microstructure occurs at significantly higher temperature than the ionic dissociation temperature. In that case, improvements in the block microphase separation occurred above the ionic dissociation temperature.

Blends of Homopolymer with Block Copolymer Ionomers [4,6]: The incorporation of specific intermolecular interactions into a blend of a homopolymer (HP) and a block copolymer (BC) increased the miscibility of the homopolymer in the block copolymer beyond the theoretical limit for athermal HP/BC blend. Relatively polar homopolymers that can interact with the sulfonate groups, e.g., poly(vinyl pyridine) and polycaprolactone, partition preferentially into the iononeric block of the block copolymer ionomer. In the case of blends of polycaprolactone (PC), which is a crystallizable polymer, with SEBS block copolymer ionomers, the PC swelled the ionic block and formed nano-sized crystallites within the ionomeric domains.

Significance of This Work: Block copolymer ionomers possess two different self-assembled microstructures that differ in size by an order of magnitude. The combination of two different microphases within a single polymer allows one considerable flexibility in the development and control of polymer morphology and properties. For example, judicious control of the ionic

at elevated temperatures superior to those of currently commercial styrene-based block copolymer TPEs. The block copolymer ionomers also combine hydrophobic and hydrophilic microdomains, which give them unique characteristics. such as the ability to swell different parts of the polymer with nonpolar or polar diluents -- including polymeric diluents. This property plus the ability to develop nano-size crystalline textures within a microdomain suggest that these materials may have application in technologies such as membranes, controlled release, solid-state batteries, interfacial agents or chemical or biological sensors. In addition, combination of the microstructures of an ionomer and a block copolymer allowed us to access for the first time the order-disorder transition of the ionic microstructure of an ionomer. Thus, these materials also allow one to pursue studies of ionomer physics that have been hitherto improssible.

Table 1: Characteristics of the block copolymers studied and summary of results

Architecture	Designation	PS content, wt% ^a	M_n , daltons ^b	Morphological Characteristics		T_{ODT} oCf
				Typec, d	Dimension, nme	
Diblock	SEP10	10.8	48200	Spheres	24.6	180
Diblock	SEP20	20.3	54000	Cylinders	40.3	325
Diblock	SEP35	35.6	52000	Lamellae	53.2	
Diblock	SEP50	50.4	52000	Lamellae	49.9	
Triblock	SEPS9	9.0	54900			80
Triblock	SEPS21	21.5	49300	Cylinders	23.2	150
Triblock	SEPS50	50.8	50000	Lamellae	30.1	275

Legend: a : Determined through proton NMR; b : Determined through gel permeation chromatography (GPC);

c : Determined by SAXS; d : Determined by TEM; e : Dimension $D = 2\pi/q^*$; f : Determined by DMTA

B. PUBLICATIONS

- Steckle, W. P., Jr., X. Lu, and R. A. Weiss. 1991. Characterization of Block Copolymer Ionomers by Small Angle x-Ray Scattering and Dynamic Mechanical Analysis. Polym. Mat. Sci. Eng., 65: 240-241.
- 2. Weiss, R. A. 1992. TPE Block Copolymer Ionomers. Proceedings of TPE'92: Sixth International Conference on Thermoplastic Elastomer Markets and Products, pp. 61-79.
- Steckle, W. P., X. Lu, and R. A. Weiss. 1992. Modification of Block Copolymer Ionomers. Polym. Preprints, 33(2): 302-303.
- 4.. Lu, X., R. A. Weiss, B. S. Hsiao, D. G. Wu, Y. J. Li, and B. Chu, 1992. Microstructure of Blends of Block Copolymer Ionomers and Poly(caprolactone) or Poly(vinylpyridene). Polym. Preprints, 33(2): 589-590.
- 5. Lu, X., W. P. Steckle and R. A. Weiss. 1993. Morphological Studies of a Block Copolymer Ionomer, Proc. An. Tech. Conf., Soc. Plast. Eng. 2842-2844.
- 6. Lu, X., and R. A. Weiss. 1993. Morphology and Phase Behavior of Blends of a Styrenic Block Copolymer Ionomer and Poly(caprolactone). Macromolecules, <u>26</u>: 3615-3622.
- 7. Lu, X., W. P. Steckle and R. A. Weiss. 1993. Ionic Aggregation in a Block Copolymer Ionomer. Macromolecules, 26: 5876-5884.
- 8. Lu, X., W. P. Steckle and R. A. Weiss. 1993. Morphological Studies of a Triblock Copolymer Ionomer by Small Angle X-Ray Scattering. Macromolecules, 26: 6525-6530.
- 9. Mani, S., R. A. Weiss, C. E. Williams, and S. F. Hahn. 1994. Characterization of the Block Microstructure in Block Copolymer Ionomers Using Small Angle X-Ray Scattering. Polym. Mat. Sci. Eng., 71: 565-566.
- 10. Lu, X., W. P. Steckle, Jr.; B. Hsiao and R. A. Weiss. 199. Thermally Induced Microstructure Transitions in a Block Copolymer Ionomer. Macromolecules, in press.
- 11. Mani, S., R. A. Weiss, S. F. Hahn, C. E. Williams, M. E. Cantino and L. H. Khairallah. 1995. Microstructure of Styrenic Block Copolymers and Their Sulfonated Ionomers. Ann. Tech. Conf., Soc. of Plast. Eng., in press.
- Mani, S., R. A. Weiss, S. F. Hahn, C. E. Williams, M. E. Cantino and L. H. Khairallah. 199 Microstructure of Block Copolymers of Polystyrene and Poly(r-ethylene-co-r-propylene). Macromolecules, submitted.
- C. Patents: None

Distribution List for Final Reports

Attach a copy of the REPORT DOCUMENTATION PAGE (DD FORM 1473) to your final report as the first page and mail two copies (including the postcard labelled DTIC FORM 50) to:

Defense Technical Information Center Building 5, Cameron Station Alexandria, VA 22314

This allows other investigators to obtain copies of your report directly from DTIC. DTIC will fill out the postcard DTIC ACCESSION NOTICE (DTIC FORM 50) and return it to you with their number for your report. When you refer people to DTIC to get a copy of your report, give this number to expedite the request.

Mail one copy to each of the following and attach this very page to the back of your report – otherwise the folks below will think they have mistakenly received a copy meant for the Molecular Biology Program):

- (a) Dr. Michael Marron ONR Code 1141 Molecular Biology Program 800 N. Quincy Street Arlington, VA 22217-5000
- (b) Administrative Grants Officer
 ONR Resident Representative
 (address varies see copy of your
 grant/contract)
- (c) Director,
 Applied Research Directorate
 ONR Code 12
 800 N. Quincy Street
 Arlington, VA 22217-5000
- (d) Director
 Office of Naval Technology
 Code 22
 800 N. Quincy Street
 Arlington, VA 22217-5000

- (e) Director
 Chemical and Biological Sci Div
 Army Research Office
 P. 0. Box 12211
 Research Triangle Park, NC 27709
- (f) Life Sciences Directorate
 Air Force Office of Scientific Res
 Bolling Air Force Base
 Washington, DC 20332
- (g) Director
 Naval Research Laboratory
 Technical Information Div
 Code 2627
 Washington, DC 20375